

## Research Paper

## Heavy Metal Contamination in the Unrefined Salt From the Kal Shur River of Gonabad City, Iran: Potential Health Risks for Local Populations

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**ABSTRACT**

**Background:** Excessive salt intake is a major public health concern due to its association with hypertension and chronic diseases. In Gonabad City, Iran, local populations consume unrefined salt from seasonal rivers, such as Kal Shur River, which may contain hazardous levels of heavy metals.

**Methods:** Twenty unrefined salt samples were collected from the Kal Shur River during the dry season. Concentrations of arsenic (As), cadmium (Cd), mercury (Hg), lead (Pb), copper (Cu), zinc (Zn), and iron (Fe) were measured using inductively coupled plasma optical emission spectroscopy. Estimated dietary intakes were calculated based on average salt consumption and compared with national standards and provisional tolerable weekly intakes (PTWIs).

**Results:** Concentrations of Pb (2.10±1.74 mg/kg; 95% CI, 1.65%, 2.55%) and Hg (0.16±0.092 mg/kg; CI, 0.14%, 0.18%) exceeded permissible limits. Weekly Pb intake reached 588% of the former PTWI, suggesting severe risk. Although As and Cd were below thresholds, their cumulative toxicity remains concerning. Essential trace elements also exceeded tolerable levels: Fe (64 mg/day), Cu (20 mg/day), and Zn (57 mg/day), raising concerns about systemic toxicity.

**Conclusion:** Unrefined salt from the Kal Shur River contains elevated levels of toxic metals, such as Pb and Hg, alongside excessive concentrations of Fe, Cu, and Zn. Occasional use may not pose an immediate risk, but regular consumption could contribute to acute and chronic health issues. These findings underscore the need for regulatory oversight and public health measures to monitor and manage heavy metal contamination in natural salt sources. Implementing environmental health policies will be critical in reducing potential risks and protecting the wellbeing of local populations.

**Keywords:** Heavy metals, Unrefined salt, Public health, Gonabad, Inductively coupled plasma optical emission spectroscopy

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## Introduction

People's health and food quality have always been closely linked. While government agencies are responsible for providing high-quality food, individuals must select and prepare their food to maintain health and prevent disease through proper hygiene practices. "Salt" is a general term for any compound formed from an acid-base reaction, also known as a neutralization reaction. It refers to sodium chloride, the crystalline substance commonly found in seawater and mineral deposits [1]. Sea and lakes are the sources of salt. Both refined and unrefined salt are edible. In the past, salt was used to add taste and keep food fresh. Salt can enter the diet in three ways: Added during food preparation, sprinkled at the table, or present in processed foods [2, 3]. According to the [World Health Organization \(WHO\)](#), adults should consume no more than 5 g of salt per day [4]. In Asia, people eat more than 12 g of salt every day, while in most other countries, people eat 9 to 12 g. People in Iran eat 10 to 15 g of salt every day, which is more than the amount suggested by the [WHO](#) [5, 6]. Hypertension is one of the most significant risk factors for cardiovascular diseases worldwide, and excessive intake of salt and sodium through the diet has a direct relationship with high blood pressure. Excessive salt consumption not only leads to raised blood pressure but also increases the risk of developing osteoporosis, colorectal cancer, and kidney stones. Reducing salt intake is considered a major public health priority in many countries [6, 7].

The presence of contaminants, such as heavy metals and contaminants, such as calcium, magnesium, and sulfate, can pose health hazards to the general public [8, 9]. Heavy metals are widely recognized as one of the most significant and well-known pollutants due to their potential to enter the human food cycle on a large scale, their extended shelf life, and their non-degradation by soil microorganisms [10]. Both natural and human-made sources contribute to the introduction of these metals into the natural cycle through soil, water, and air [11]. They also inhibit the absorption of micronutrients [12]. Arsenic (As), lead (Pb), mercury (Hg), and cadmium (Cd) are among the heavy metals of health significance [13] and their effects are discussed below:

The metal As exists in many organic and inorganic forms with varying degrees of toxicity and carcinogenic potential depending on their oxidation states. Kidney damage results from the binding of the 95–99% of absorbed As to the hemoglobin component of erythrocytes [14].

The metal Pb is soft, bluish-white, and highly toxic. Its adverse consequences are exacerbated by its environmental persistence. Among the hazardous heavy metals, Pb possesses the lowest permissible limit, indicating that even minimal concentrations in water and soil can be detrimental. Anemia, increased blood pressure, reproductive disorders, impaired vitamin D metabolism, and neurological disorders are all induced by Pb exposure in the human body [15].

Hg is an environmental pollutant of global significance and is extensively dispersed. Inorganic Hg is converted to methylmercury in lakes and wetlands by living organisms or other processes. Hg poisoning can cause headache, lightheadedness, agitation, tremors, weight loss, restlessness, depression, heart failure, bloody stools, vision problems, severe skin infections, respiratory irritation, and ultimately death [16].

Cd is another highly toxic metal for humans and is produced as a byproduct of zinc (Zn) refining. Cd also accumulates in the human body, resulting in erythrocyte destruction, nausea, excessive salivation, muscle contractions, renal destruction, chronic lung issues, and skeletal deformities [17].

Trace heavy metals (Zn, iron [Fe], copper [Cu]) play a crucial role in body due to both essential functions and their potential toxicity. Excessive intake of Zn, Cu, and Fe can lead to toxicity [18]. Most trace elements present in the human body are primarily absorbed from dietary sources [19, 20].

The Kal Shur River is the only seasonal river in Gonabad City, Iran, and in recent years, it has often remained dry due to the construction of a dam along its course and reduced rainfall. Additionally, the region's desert climate and saline soil cause significant salt accumulation in the river's water pools during the summer. This salt is collected and used by local residents. Despite the widespread use of this salt by local communities and the well-documented health risks associated with heavy metal contamination in edible salts, no prior study has systematically investigated the presence and concentration of heavy metals in salt extracted from the Kal Shur River. This represents a significant gap in environmental and public health research, particularly given the potential exposure risks to the local population who regularly consume this salt.

Therefore, this study aimed to assess the heavy metal contamination levels in salt from the Kal Shur River and calculate the estimated weekly intake of these metals

among Gonabad City residents. The findings provide essential data for public health risk assessment and inform appropriate safety guidelines for local salt consumption.

## Methods

### Study area

The Kal Shur River, which originates from the northern region of Torbat Heydarieh City, is situated 35 kilometers from Gonabad City [21]. Its drainage basin area is 12713 square kilometers, and its length is 258 kilometers [22]. Kal Shur eventually flows into the saline plain of Bejastan [23]. Salt samples were collected from accessible sections of the Kal Shur River, specifically from areas commonly used by local people for salt gathering (near the road). Figure 1 shows an aerial map of Kal Shur River in Gonabad City and its geographical location within the county.

### Sampling

Twenty unrefined salt samples were obtained from the Kal Shur River in the summer, when the river was dry. Surface samples were collected from each sampling point using wooden scoops to an approximate depth of 8 centimeters. Sample size was determined using the population mean calculation formula (Equation 1) with a 95% confidence level (CI), based on the study by Sabet et al. [11]. The calculation was performed for various heavy metals, including As, Cd, and Pb:

$$1. n = (z_{1-\alpha/2})^2 \times s^2 / d^2 = (1.96)^2 \times (0.088)^2 / (0.05)^2 = 12$$

Where:

$$z_{1-\alpha/2} = 1.96 \text{ (95\% CI)}$$

$$s = 0.088 \text{ (standard deviation from reference study)}$$

$$d = 0.05 \text{ (desired precision)}$$

The calculation indicated that As required the largest sample size among the studied metals. To ensure adequate statistical power and better representation of the study area, 20 unrefined salt samples were ultimately collected.

### Sample preparation and analysis

After labeling, samples were individually dried in a furnace at 105 °C for approximately 2 hours. After drying, each sample was separately ground and homogenized to ensure uniformity while maintaining individual sample integrity.

The Pb, Cd, Hg, As, Cu, Zn, and Fe contents of all samples were analyzed using an inductively coupled plasma optical emission spectroscopy system (Spectro Arcos-76004555 plasma/AMETEK ARCOS FHE12, Germany) at the Central Laboratory of Ferdowsi University of Mashhad, Iran. Analysis was performed by a certified analytical laboratory using standard protocols.

### Estimation of dietary exposure to heavy metals

The dietary intake of the studied heavy metals was estimated, and associated health risks were assessed by comparing results with the provisional tolerable weekly intakes (PTWIs). Exposure was calculated per kilogram of body weight by dividing the total dietary exposure by the average body weight of the population [13, 24].

The daily intake of heavy metals was determined using the Equation 2:

$$2. \text{Daily intake of heavy metals} = \sum (\text{Concentration of heavy metals in salt samples} \times \text{Mean salt intake (g/person/day)}) [25].$$

The weekly intake of heavy metals was then calculated using the following Equation 3:

$$3. \text{Weekly intake of heavy metals} = \text{Daily intake} \times 7$$

To determine the weekly intake per kilogram of body weight (for comparison with PTWIs), the Equation 4 was used:

$$4. \text{Weekly intake per kilogram of body weight (kg)} = \text{Weekly intake} / \text{Reference body weight (60 kg)} [26].$$

### Statistical analysis

All data were analyzed using SPSS software, version 21. One-sample t-test was used to determine the difference between the levels of heavy metals in the salt samples and the maximum permitted levels in the standard.  $P < 0.05$  was considered statistically significance. The correlation between metals was examined using Pearson's correlation coefficient.

## Results

Table 1 presents the concentrations of the target metals in the salt samples alongside their comparison with established standards, while Tables 2 and 3 summarize the population's dietary exposure to heavy metals through salt consumption.

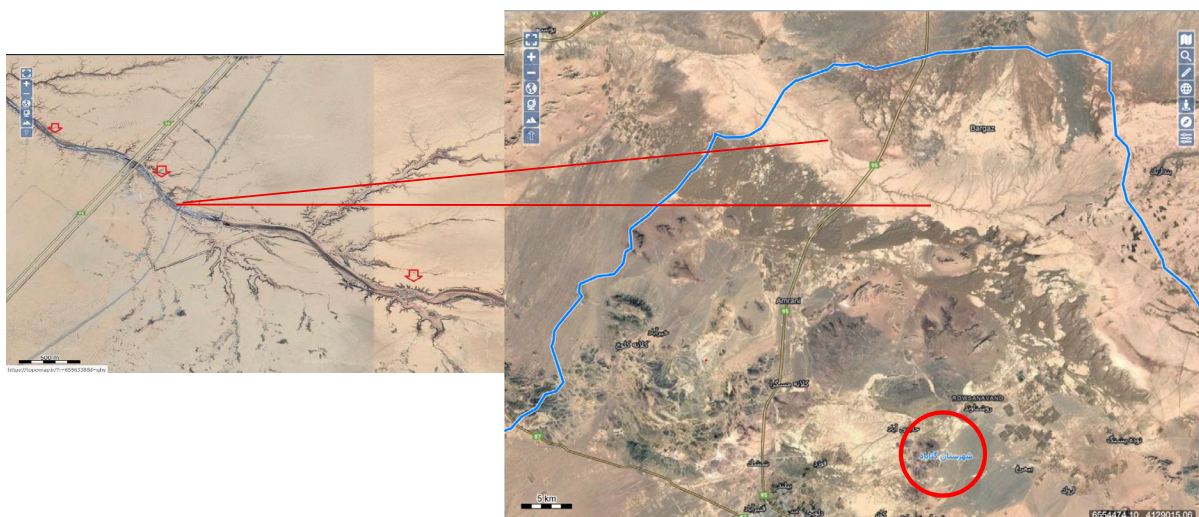


Figure 1. Location of Kal Shur in Gonabad City



The average concentration of heavy metals was Fe>Zn>Cu>Pb>As>Hg>Cd. The Mean±SE was 6.41±6.27 for Fe, 5.70±2.73 for Zn, 2.03±1.52 for Cu, 2.10±1.74 for Pb, 0.25±0.18 for As, 0.92±0.16 for Hg, 0.019±0.017 mg/kg for Cd. Among the studied salt samples, Fe had the highest and Cd had the lowest average concentration. The average concentration of As and Cd (0.25 and 0.019 mg/kg) was lower than the permissible values recommended by Iran Standard 26 (0.5 and 0.2 mg/kg, respectively). The results show that the concentrations of heavy metals Pb and Hg are higher than the standard. The levels of heavy metals Zn, Fe, and Cu exceeded the standard value, although the increase was not statistically significant for Cu. The correlation analysis showed several significant associations among

the studied metals. The strongest was between As and Cu (r=0.837, P<0.01), followed by Cu and Fe (r=0.602, P<0.01), and Zn with both Cu (r=0.544, P<0.01) and As (r=0.663, P<0.01). These relationships likely reflect shared geochemical processes, such as dissolution and sedimentation, which may have simultaneously introduced these elements into the river. Pb also exhibited significant positive correlations with Zn (r=0.405, P<0.01) and As (r=0.282, P<0.05), suggesting possible common anthropogenic or environmental sources. In contrast, Cd showed no significant associations and, in some cases, negative correlations, indicating a distinct geochemical behavior. The correlations between the metals were assessed using Pearson’s correlation coefficient, although the table is not shown here.

Table 1. Mean concentration of heavy metals in Kal Shur’s salt

Metal	Mean±SD (ppm)	Standard Value	P**
As	0.18±0.25	0.5*	<0.001
Cd	0.017±0.019	0.2*	<0.001
Hg	0.092±0.16	0.05*	<0.001
Pb	1.7±2.1	1*	<0.001
Cu	1.52±2.03	2	0.844
Fe	6.27±6.41	10	<0.001
Zn	2.73±5.7	0.5	<0.001



Abbreviations: As: Arsenic; Cd: Cadmium; Hg: Mercury; Pb: Lead; Cu: Copper; Zn: Zinc; Fe: Iron; INSO: Food and feed-maximum limit of heavy metals.

\*Standard values based on INSO 26, \*\*A one-sample t-test.

**Table 2.** Dietary exposure of consuming Kal Shur's salt to heavy metals

Metal	Acceptable Daily Intake (mg/day)	PTWI	Weekly Intake (mg/kg)	PTWI (%)
As	-	0.015*	0.00175	11.6
Cd	1	0.007	0.00133	19
Hg	0.4	0.004	0.00112	28
Pb	-	0.025	0.147	588

Abbreviations: PTWI: Provisional tolerable weekly intakes; As: Arsenic; Cd: Cadmium; Hg: Mercury; Pb: Lead; JECFA: Joint FAO/WHO expert committee on food additives. 

\* JECFA has withdrawn PTWI for Pb and As due to their toxicity, concluding that no safe intake level can be established (previous PTWI).

The results of this study indicate that the concentrations of several heavy metals in unrefined salt from the Kal Shur River exceed established health-based limits, posing potential risks to consumers. According to Table 2, the weekly intake of Pb and Hg notably surpassed their respective PTWI values, with Pb intake reaching 588% of the previously established tolerable level, indicating a significant health hazard. While As and Cd were below standard thresholds, their presence—even at low levels—should not be overlooked due to their cumulative toxicity.

While essential elements such as Fe, Cu, and Zn are vital for various biological functions and contribute positively to health, they can cause toxicity at high concentrations. These elements serve critical roles, such as enzyme co-factor processing, immune enhancement, and oxygen transport. In contrast, toxic metals, such as Pb and Hg have no beneficial role in the body, and even small exposures can be hazardous. The key difference lies in the potential for accumulation and the lack of a physiological role for these toxic metals, as opposed to the threshold-dependent toxicity of essential elements.

Furthermore, as shown in Table 3, the daily intake of essential trace elements—particularly Fe, Cu, and Zn—were found to greatly exceed the upper intake levels (ULs) established for adults. Fe intake was calculated at 64 mg/day, which is over 1.4 times the recommended UL (45 mg/day), followed by Cu at 20 mg/day (UL=10 mg/day) and Zn at 57 mg/day (UL=40 mg/day). These excessive intakes can potentially lead to systemic toxicity, affecting the liver, kidneys, gastrointestinal tract, and metabolic functions.

## Discussion

The findings of this study show that the concentration of heavy metals in unrefined salts followed the order Fe>Zn>Cu>Pb>Hg>As>Cd. Among these, Fe showed the highest average level (6.41±6.27 mg/kg), while Cd had the lowest (0.019±0.017 mg/kg). Compared to the national standard limits set by ISIRI, the concentrations of Pb and Hg exceeded permissible values, while As and Cd remained within acceptable thresholds. However, even low levels of Cd and As pose cumulative health risks such as nephrotoxicity, carcinogenicity, and developmental effects [27, 28].

**Table 3.** The comparison of obtained metal values to maximum permissible limits

Metal	RDA/AI (mg/day)	Upper Limit	The Obtained Value (mg/day)
Cu	0.9	10	20
Fe	8-18*	45	64
Zn	8-11**	40	57

Abbreviations: Cu: Copper; Zn: Zinc; Fe: Iron; RDA: Recommended dietary allowance; AI: Adequate intakes. 

\*For adults ages 31-50, the recommended values apply to men and women, respectively. For individuals over 50 years old, 8 mg is advised for both men and women; \*\*For women and men, respectively.

In contrast to our findings, Pourgheysari et al. (2015) reported lower mean concentrations of Pb (1.38 mg/kg), Hg (0.66 mg/kg), and Fe (18.12 mg/kg) in salt samples from Isfahan, indicating a significant regional or source-based variation [13]. Similarly, Heshmati et al. (2014) observed Cd levels below 0.01 mg/kg, which is considerably lower than our measured concentration, although both remain within safe limits [29]. Such discrepancies might be attributed to differences in refining methods, geological sources, and environmental pollution.

Zn and Cu, although essential trace elements, were found in our study at  $5.70 \pm 2.73$  mg/kg and  $2.03 \pm 1.52$  mg/kg, respectively. These values are consistent with those reported by Khaniki et al. (2007), who recorded Zn and Cu concentrations of 6.1 and 3.5 mg/kg, respectively, in Tehran market salts [30]. While these levels may contribute positively to daily nutrient intake, chronic exposure above tolerable ULs—particularly Cu (>10 mg/day) and Zn (>40 mg/day)—can lead to gastrointestinal distress, hepatic damage, and impaired immune function [31, 32].

The significantly elevated Fe concentration found in our study (64 mg/day estimated intake) far surpasses the established UL of 45 mg/day, suggesting a potential risk of Fe overload disorders, such as hemochromatosis or oxidative stress-related tissue damage [33, 34]. In comparison, Foroughi and Ebrahimpour (2021) reported lower levels of Fe, Zn, and Pb in salts from Maharloo Lake and Korsia mines, reinforcing the idea that the mineral composition varies significantly by geographical origin and source [20].

Our findings are further supported by Shariatifar et al. (2021), who found higher levels of Pb and Zn in unrefined salts from Urmia compared to refined samples, emphasizing the role of purification processes [35]. Similarly, Kokabi and Pakhirehzan (2024) concluded that traditional and seaweed salts contain higher mineral and metal loads than industrially refined salts, which may account for the elevated levels seen in our unrefined samples [36]. Geological factors play a pivotal role in the observed variations in heavy metal concentrations. Salts mined from areas with natural heavy metal ores, such as those in the Māhnehān Mountains of northwestern Iran, including the Chehrābād salt mine, are likely to contain elevated levels of Fe, Zn, Pb, and Cu due to the region's mineral-rich geological profile [37]. Similarly, Di Salvo et al. (2023) investigated gourmet salts from various global regions and found that Persian Blue salt from Iran exhibited notably high Fe (21.72 mg/kg) and Zn (1334.44 mg/kg) concentrations, attributed to the

Fe- and calcium-rich geological formations of the mining areas [38]. These studies confirm that the geological context is a primary driver of heavy metal content in unrefined salts, explaining the elevated levels observed in our samples compared to those from other regions.

Environmental and industrial contributions to heavy metal accumulation must also be considered. Industrial activities associated with salt extraction and processing further contribute to heavy metal contamination, though their impact is secondary to geological factors for unrefined salts. Mining operations, particularly in regions like Chehrābād, often involve heavy machinery and explosives, which can disturb geological layers and introduce metals such as chromium, Pb, or nickel into the extracted salt [37]. According to Liu et al. (2020), industrial activities and atmospheric deposition are major contributors to heavy metal presence in soils and surface waters, which in turn affect salt sources [39]. In contrast, Cheraghali et al. (2010) found that heavy metal levels in Iranian rock salts did not significantly differ between unrefined and refined samples, suggesting that the inherent geological composition overshadows industrial contributions in unrefined salts [34]. However, proximity to industrial or agricultural operations can exacerbate pollution, especially in sea salts, where pollutants from industrial effluents or agricultural runoff may accumulate in seawater.

This is consistent with Kalteh et al. (2020), who identified unsafe levels of metal fumes in workers of salt production facilities, suggesting occupational and environmental exposure risks [40].

Furthermore, excessive intake of salts contaminated with heavy metals poses substantial public health risks. Azadnajafabad et al. (2021) demonstrated that Iran has one of the highest per capita salt consumption rates, amplifying the health implications of even moderate contamination [41]. Chronic exposure to Pb and Hg, both found above standard limits in our samples, is associated with neurotoxicity, renal impairment, and reproductive dysfunction [27, 31]. As emphasized by Baseri et al. (2021), dietary exposure through staple foods, including cheese and salt, significantly contributes to the total heavy metal burden in the Iranian population [19].

Our results confirm that both essential and toxic metals are present in edible salts at varying concentrations, some of which exceed national or international health-based thresholds. These findings are consistent with prior national and international studies and point toward the need for more stringent monitoring and refinement

practices. Public education and regulatory interventions are also vital to mitigate the health risks posed by long-term consumption of contaminated salt.

## Conclusion

This study examined the current levels of heavy metal concentrations in salt samples collected from the Kal Shur region. The findings revealed that certain metals, particularly Pb and Hg, were present at concentrations exceeding permissible limits, raising significant concerns about their potential impact on public health, especially regarding adverse effects on the nervous and renal systems. Continuous exposure, even at low levels, may lead to serious health issues and poses greater risks to vulnerable populations, such as children and pregnant women. Therefore, enhancing salt purification processes, implementing rigorous quality control measures, and developing hygiene education programs, particularly targeting these sensitive groups, are essential. Additionally, heavy metals can enter the human body through other dietary sources, including meat and its products, poultry, fish, and drinking water. Given their cumulative nature, it is critical to assess the concentrations of these metals in other sources, which requires updated data on dietary consumption patterns in Iran. As an initial step in identifying the current situation, this study underscores the need for further research to better understand the biological effects of these contaminants and to develop management strategies aimed at reducing potential health risks and promoting public health through preventive measures.

## Ethical Considerations

### Compliance with ethical guidelines

This study was approved by [Gonabad University of Medical Sciences](#), Gonabad, Iran (Code: IR.GMU.REC.1399.109).

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### Authors' contributions

Conceptualization, editing, and literature review: Zohreh Abdi-Moghadam and Davoud Salarbashi; Methodology: Zohreh Abdi-Moghadam and Nasim Khajavian; Investigation, supervision and data collection:

Zohreh Abdi-Moghadam; Writing the original draft: Zohreh Abdi-Moghadam and Nazanin Tayebi.

### Conflict of interest

The authors declared no conflicts of interest.

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